

Contents List available at JACS Directory

# Journal of Advanced Chemical Sciences

journal homepage: www.jacsdirectory.com/jacs



# Quinazoline Based Synthesis of Some Novel Heterocyclic Schiff Bases

V. Navale<sup>1</sup>, R. Shinde<sup>1</sup>, S. Patil<sup>2</sup>, A. Vibhute<sup>3</sup>, S. Zangade<sup>4,\*</sup>

<sup>1</sup>Department of Chemistry, Dayanand Science College, Latur – 413531, Maharashtra, India. <sup>2</sup>Department of Chemistry, Maharashtra Udaigiri Mahavidyalaya, Udgiri – 413517, Maharashtra, India.

<sup>3</sup> Department of Chemistry, Priyadarshini Indira Gandhi Engineering College, Nagpur – 440019, Maharashtra, India.

<sup>4</sup>Department of Chemistry, Madhavrao Patil College, Palam, Dist. Parbhani – 431720, Maharashtra, India.

### ARTICLE DETAILS

## A B S T R A C T

NMR, IR and mass spectroscopy.

Article history: Received 11 December 2015 Accepted 23 December 2015 Available online 24 December 2015

*Keywords:* Heterocyclic Schiff Bases Quinazolines Acetophenones

#### 1. Introduction

Quinazoline is a compound made up of two fused six-membered simple aromatic rings, structure compound containing benzene fused to pyrimidine. Medicinally it has been used in various areas as an analgesic and anti-inflammatory [1-4], antihypertensive [5, 6], antimicrobial [7-9], antibacterial [10], anticonvulsant [11-13], anticancer [14, 15], antimalarial [16] and antidepressant activities [17].

Schiff bases (imines) are well known for their wide applications and are useful intermediates in organic synthesis [18]. These compounds have intrinsic biological activities including anticancer [19], antitumour [20], antitubercular [21], antibacterial [22], antioxidant [18], and antiproliferative [23] activities. Moreover, Schiff bases also exhibit fluorescence [24], photoluminescence [25], and aggregation [26] properties. In view of these observations, we plan to synthesize some novel quinazoline based imines **3a-i** by a condensation reaction of a substituted acetophenones with 3-amino-2-methyl-3H-quinazolin-4-one in presence of acetic acid using classical procedure (Fig. 1).

#### 2. Experimental Methods

#### 2.1 Chemistry

Melting points were determined in open capillary tubes and are uncorrected. FT-IR spectra were recorded in KBr pellets on a Perkin-Elmer [8201] spectrometer. <sup>1</sup>H NMR spectra were recorded on a Gemini 300-MHz instrument in DMSO-d<sub>6</sub> as the solvent and TMS was used as an internal standard. The mass spectra were recorded on SHIMADZU (GCMS-QP 1000 EX) GC-EI-MS spectrometer. Elemental analyses were performed on a Perkin-Elmer 240 CHN elemental analyser. Purification of the compound was indicated using TLC (ethyl acetate / cyclohexane (0.25 mL: 0.25 mL, v/v) as the mobile phase).

### 2.1.1 Preparation of Schiff bases

In 50 mL of round bottom flask, mixture of 3-amino-2-methyl-3Hquinazolin-4-one **1** (0.01mole) and substituted acetophenones **2a-i** (0.01 mole) was dissolved in methyl alcohol (15 mL). To this reaction mixture

\*Corresponding Author Email Address: drsbz@rediffmail.com (S. Zangade)

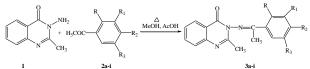
2394-5311 / JACS Directory©2015-2016. All Rights Reserved

3-Amino-2-methyl-3H-quinazolin-4-one on condensation with different substituted acetophenones in

presence of acetic acid under classical procedure to affords novel series of Schiff bases containing

quanzoline moiety. The newly synthesized imines are confirmed on the basis of spectral technique, <sup>1</sup>H

acetic acid (0.001 mole) was added and resultant reaction mixture was refluxed for 2 to 3 hrs. On completion of reaction as monitored by TLC (ethyl acetate: cyclohexane, 0.25 mL: 0.25 mL, v/v as the mobile phase) the reaction mixture was work-up using cold water to obtained crud solid product. The separated solid was filtered and recrystallised from ethanol to yield pure Schiff's bases **3a-i**. Analytical and physical data are given in table 1.



**3a**: R, Cl; R<sub>1</sub>, Cl; R<sub>2</sub>, Cl; R<sub>3</sub>, H **3b**: R, OH; R<sub>1</sub>, Br; R<sub>2</sub>, OH; R<sub>3</sub>, Br **3c**: R, H; R<sub>1</sub>, Cl; R<sub>2</sub>, Cl; R<sub>3</sub>, H **3d**: R, Cl; R<sub>1</sub>, H; R<sub>2</sub>, Cl; R<sub>3</sub>, H **3e**: R, OH; R<sub>1</sub>, H; R<sub>2</sub>, CH<sub>3</sub>; R<sub>3</sub>, Cl **3f**: R, OH; R<sub>1</sub>, Br; R<sub>2</sub>, H; R<sub>3</sub>, Cl **3g**: R, OH; R<sub>1</sub>, Br; R<sub>2</sub>, H; R<sub>3</sub>, CH<sub>3</sub> **3h**: R, H; R<sub>1</sub>, H; R<sub>2</sub>, OCH<sub>3</sub>; R<sub>3</sub>, H **3i**: R, H; R<sub>1</sub>, H; R<sub>2</sub>, Cl; R<sub>3</sub>, H

Fig. 1 Synthesis of some novel heterocyclic Schiff bases

2-Methyl-3-[1-(2,3,4-trichloro-phenyl)-ethylideneamino]-3H-quinazolin-4one (3a)

Appearance: Light Yellow Solid, FT-IR (KBr, ν, cm<sup>-1</sup>): 3228, 3066, 2945, 1788, 1618, 1547, 1454, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, δ, ppm): 6.72-8.32 (m, 6H, ArH), 2.60 (s, 3H, CH<sub>3</sub>), 2.71 (s, 3H, CH<sub>3</sub>). (MS (EI), *m/z* (%): 380.5 (M+, 62%).

3-[1-(3,5-Dibromo-2,4-dihydroxy-phenyl)-ethylideneamino]-2-methyl-3Hquinazolin-4-one (3b)

Appearance: Light Yellow Solid, FT-IR (KBr, ν, cm<sup>-1</sup>): 3225, 3087, 2950, 1786, 1620, 1542, 1450, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, δ, ppm): 12.1 (s, 2H, OH), 6.69-8.29 (m, 5H, ArH), 2.61 (s, 3H, CH<sub>3</sub>), 2.73 (s, 3H, CH<sub>3</sub>). (MS (EI), *m/z* (%): 467 (M+, 68%).

3-[1-(3,4-Dichloro-phenyl)-ethylideneamino]-2-methyl-3H-quinazolin-4one (3c)

Appearance: Light Yellow Solid, FT-IR (KBr, ν, cm<sup>-1</sup>): 3229, 3090, 2953, 1784, 1620, 1558, 1457, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, δ, ppm): 6.70-8.26 (m, 7H, ArH), 2.60 (s, 3H, CH<sub>3</sub>), 2.72 (s, 3H, CH<sub>3</sub>). (MS (EI), *m/z* (%): 346 (M+, 70%).

Cite this Article as: V. Navale, R. Shinde, S. Patil, A. Vibhute, S. Zangade, Quinazoline based synthesis of some novel heterocyclic Schiff bases, J. Adv. Chem. Sci. 2(1) (2016) 201-203.

3-[1-(2,4-Dichloro-phenyl)-ethylideneamino]-2-methyl-3H-quinazolin-4one (3d)

Appearance: Light Yellow Solid, FT-IR (KBr, ν, cm<sup>-1</sup>): 3235, 3087, 2960, 1786, 1620, 1562, 1450, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, δ, ppm): 6.73-8.29 (m, 7H, ArH), 2.60 (s, 3H, CH<sub>3</sub>), 2.73 (s, 3H, CH<sub>3</sub>). (MS (EI), *m/z* (%): 346.5 (M+, 50%).

#### 3-[1-(5-Chloro-2-hydroxy-4-methyl-phenyl)-ethylideneamino]-2-methyl-3H-quinazolin-4-one (3e)

Appearance: Light Yellow Solid, FT-IR (KBr, v, cm<sup>-1</sup>): 3230, 3082, 2937, 1785, 1620, 1550, 1431, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ,  $\delta$ , ppm): 12.10 (s, 1H, OH), 6.78-8.25 (m, 6H, ArH), 2.31 (s, 3H, CH<sub>3</sub>), 2.60 (s, 3H, CH<sub>3</sub>), 2.73 (s, 3H, CH<sub>3</sub>). (MS (EI), m/z (%): 341.5 (M+, 43%).

#### 3-[1-(3-Bromo-5-chloro-2-hydroxy-phenyl)-ethylideneamino]-2-methyl-3Hquinazolin-4-one (3f)

Appearance: Light Yellow Solid, FT-IR (KBr, ν, cm<sup>-1</sup>): 3234, 3067, 2981, 1788, 1620, 1527, 1447, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, δ, ppm): 12.10 (s, 1H, OH), 6.74-8.28 (m, 6H, ArH), 2.60 (s, 3H, CH<sub>3</sub>), 2.72 (s, 3H, CH<sub>3</sub>). (MS (EI), *m/z* (%): 406.5 (M+, 75%).

#### 3-[1-(3-Bromo-2-hydroxy-5-methyl-phenyl)-ethylideneamino]-2-methyl-3H-quinazolin-4-one (3g)

Appearance: Light Yellow Solid, FT-IR (KBr, ν, cm<sup>-1</sup>): 3230, 3047, 2970, 1787, 1620, 1530, 1450, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, δ, ppm): 12.10 (s, 1H, OH), 6.68-8.23 (m, 6H, ArH), 2.31 (s, 3H, CH<sub>3</sub>), 2.61 (s, 3H, CH<sub>3</sub>), 2.73 (s, 3H, CH<sub>3</sub>). (MS (EI), *m/z* (%): 386 (M+, 65%).

3-[1-(4-Methoxy-phenyl)-ethylideneamino]-2-methyl-3H-quinazolin-4-one (3h)

Appearance: Light Yellow Solid, FT-IR (KBr, ν, cm<sup>-1</sup>): 3228, 3061, 2923, 1786, 1620, 1568, 1416, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, δ, ppm): 6.70-8.26 (m, 8H, ArH), 3.83 (s, 3H, OCH<sub>3</sub>), 2.61 (s, 3H, CH<sub>3</sub>), 2.73 (s, 3H, CH<sub>3</sub>). (MS (EI), *m/z* (%): 307 (M+, 40%).

# 3-[1-(4-Chloro-phenyl)-ethylideneamino]-2-methyl-3H-quinazolin-4-one (3i)

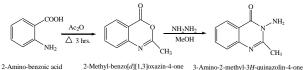
Appearance: Light Yellow Solid, FT-IR (KBr, ν, cm<sup>-1</sup>): 3230, 3042, 2938, 1786, 1620, 1554, 1422, 810, 650. <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, δ, ppm): 6.74-8.29 (m, 8H, ArH), 2.62 (s, 3H, CH<sub>3</sub>), 2.72 (s, 3H, CH<sub>3</sub>). (MS (EI), *m/z* (%): 311.5 (M+, 80%).

Table 1 Analytical and physical data of compounds (3a	- i	ij	)
---	-----	----	---

Entry	Mol. Formula	Yield	M.P.	Elemental Analysis found [calculated]			
			<sup>0</sup> C	С	Н	X=Cl, Br,I	N
3a	$C_{17}H_{12}ON_3Cl_3$	76	64	53.43	3.25	27.70	11.25
				[53.61]	[3.15]	[27.98]	[11.03]
3b	$C_{17}H_{13}O_3N_3Br_2$	69	179	43.59	2.84	34.33	8.89
				[43.68]	[2.78]	[34.26]	[8.99]
3c	$C_{17}H_{13}ON_3Cl_2$	78	70	58.77	3.60	20.45	12.50
				[58.95]	[3.75]	[20.52]	[12.13]
3d	$C_{17}H_{13}ON_3Cl_2$	75	142	58.77	3.80	20.58	12.16
				[58.95]	[3.75]	[20.52]	[12.13]
3e	$C_{18}H_{16}O_2ClN_3$	68	110	63.11	4.60	10.82	12.34
				[63.25]	[4.68]	[10.76]	[12.29]
3f	$C_{17}H_{13}O_2N_3BrCl$	72	172	50.35	3.33	10.92	8.82
				[50.18]	[3.19]	[10.33]	[8.73]
3g	$C_{18}H_{16}O_2N_3Br$	70	190	55.60	4.17	10.60	20.61
				[55.95]	[4.14]	[10.88]	[20.72]
3h	$C_{18}H_{17}O_2N_3$	60	210	70.15	5.15	-	13.2
				[70.35]	[5.53]		[13.6]
3i	$C_{17}H_{14}ON_3Cl$	76	185	65.11	4.41	13.52	11.15
				[65.48]	[4.49]	[13.48]	[11.39]

#### 3. Result and Discussion

In view of the biological and medicinal importance of imines (Schiff bases), the present studies describe synthesis of novel quinazoline based synthesis of Schiff bases **3a-i**. The starting 3-amino-2-methyl-3H-quinazolin-4-one **1** were prepared by the action of 2-amino benzoic acid with acetic anhydride to yield 2-Methyl-benzo[d][1,3]oxazin-4-one, which on further condensation with hydrazine hydrate to give 3-amino-2-methyl-3H-quinazolin-4-one (Fig. 2). The obtained **1** on condensation with substituted acetophenones in presence of acetic acid using classical route to obtained imines **3a-i**. The use classical procedure for synthesis of imines is found to efficient in term of short reaction time, operationally being simple better yield of products and reactions were take place smoothly without any typical catalyst and modified apparatus.



2-Amino-benzoic acid 2-Meuryr-benzolaj(1,5 joxazin-4-bie 5-Amino-2-methyl-5H-quinazoin-4-

Fig. 2 Synthesis of 3-amino-2-methyl-3H-quinazolin-4-one

The structures of newly synthesized imines **3a-i** were confirmed on the basis of spectral analysis. In IR spectra of condensed product display the disappearance absorption band due C=O stretch and appearance of band near 1620 cm<sup>-1</sup> due to C=N stretch of azomethine. The absorption band observed near 3230 cm<sup>-1</sup> is due to presence of OH stretch. In <sup>1</sup>H NMR spectrum of imines display near  $\delta$  2.73 ppm due to singlet of CH<sub>3</sub> of quinazolines nucleus and at  $\delta$  2.60 ppm due to CH<sub>3</sub> of imines. The singlet of hydroxy proton apperas near  $\delta$  12.1 ppm. Therefore all synthesized compounds exhibited satisfactory spectral data consistent with their structures.

#### 4. Conclusion

In summary, we have synthesized novel series of quinazolines based Schiff bases using classical procedure under slightly acidic conditions.

#### References

- C.H. Rajveer, C.H. Swarnalatha, R.B. Stephen, S. Sudhrshini, Synthesis of 6bromo-oxo quinazoline derivatives and their pharamcological activities, Int. J. Chem. Res. 1 (2010) 21-24.
- [2] A.M. Alafeefy, A.A. Kadi, O.A. Al-Deeb, Synthesis, analgesic and antiinflammatory evaluation of some novel quinazoline derivatives, Eur. J. Med. Chem. 45 (2010) 4947-4952.
- [3] P.S. Salunkhe, H.M. Patel, R.D. Shimpi, N.N. Lalwani, Study of analgesic & antiinflammatory evaluation of some 2,3- dihydroquinozoline-4-one derivatives, Int. J. Pharm. Res. 2 (2010) 974-987.
- [4] M.V. Aanandhi, V. Velmurugan, S. Shanmugapriya, Synthesis, characterization and biological evaluation of 3,4-Dihydro quinazoline2(H)-one derivatives, J. Chem. Pharm. Res. 2 (2011) 676-683.
- [5] H.U. Patel, R.S. Patel, C.N. Patel, Synthesis and antihypertensive activity of some quinazoline derivatives, J. Appl. Pharma. Sci. 3 (2013) 171-174.
- [6] E. Honkane, P. Kairisalo, P. Nore, H. Karppanen, Synthesis and antihypertensive activity of some new quinazoline derivatives, J. Med. Chem. 10 (1983) 1433-1438.
- [7] A.A. Aly, Synthesis and antimicrobial activity of some annelated quinazoline derivatives, J. Chin. Chem. 54 (2007) 437-446.
- [8] T.P. Selvam, P.V. Kumar, C. Rajaram, Synthesis and antimicrobial activity of novel thiazoloquinazoline derivatives, Int. J. Pharma. Sci. 2 (2010) 119-122.
- [9] L.M. Break, M.A. Mosselhi, Synthesis, structure and antimicrobial activity of new 3 and 2-arylmethyl and arylacyl-3H[1,2,4]triazino[3,2-b]-quinazoline-2,6(1H) diones as expected as DNA fluorphores, J. Chem. Sci. 5 (2012) 23-28.
- [10] B. Kiruthiga, K. Ilango, P. Valentina, Synthesis of some new 2-substituted quinazolin-4-one derivatives and their biological activities, Int. J. Pharm. Tech. Res. 4 (2009) 1503-1506.
- [11] A.G. El-helby, M.H. Abdel wahab, Design and synthesis of some new derivatives of 3H- quinazolin-4-one with promising anticonvulsant activity, Acta Pharm. 53 (2003) 127-138.
- [12] P. Ilangovan, S. Ganguly, V. Pandi, J.P. Stables, Design and synthesis of novel quinazolinone derivatives as broad spectrum anticonvulsants, Sch. Res. Lib. 1 (2010) 13-21.
- [13] A.A. Kadi, A.S. El-Azab, A.M. Alafeefy, Synthesis and biological screening of some new substituted 2-mercapto-4-(H)-quinazolinone analogs as anticonvulsant agents, Az. J. Pharm. Sci. 34 (2008) 135-155.
- [14] A.G. Nerkar, A.K. Saxena, S.A. Ghone, In silico screening, synthesis and *In vitro* evaluation of some quinazolinone and pyridine derivatives as dihydrofolate reductase inhibitors for anticancer activity, J. Chem. 6 (2009) 97-102.
- [15] K. Manasa, R.V. Sidhaye, G. Radhika, C.N. Nalini, Synthesis, antioxidant and anticancer activity of quinazoline derivatives, Curr. Pharma Res. 2 (2011) 101-105.
- [16] J. Guan, Q. Zhang, M. O'Neil, Antimalarial activities of new pyrrolo[3,2f]quinazoline-1,3-diamine derivatives, Am. Soc. Micro. 12 (2005) 4928–4933.
- [17] H.J. Wang, C.X. Wei, X.Q. Deng, Synthesis and evaluation on anticonvulsant and antidepressant activities of 5-Alkoxy tetrazolo[1,5-a]quinazolines, Arch. Pharm. 11 (2009) 671-675.
- [18] S. Zangade, A. Shinde, S. Chavan, Y. Vibhute, Solvent-free, environmentally benign syntheses of some imines and antioxidant activity, Orbital: Electron. J. Chem. 7 (2015) 208-214.
- [19] B.S. Creaven, B. Duff, D.A. Egan, K. Kavanagh, G. Rosair, V. Reddy Thangell, M. Walsh, Anticancer and antifungal activity of copper(II) complexes of quinolin-2(1*H*)-one-derived Schiff bases, Inorg. Chim. Acta 363, (2010) 4048-4058.
- [20] G. Hu, G. Wang, N. Duan, X. Wen, T. Cao, S. Xie, W. Huang, Design, synthesis and antitumor activities of fluoroquinolone C-3 heterocycles (IV): s-triazole Schiff-Mannich bases derived from ofloxacin, Acta Pharm. Sin. B. 2 (2012) 312-317.
- [21] T. Aboul-Fadl, F.A.S. Bin-Jubair, O. Aboul-Wafa, Schiff bases of indoline-2,3dione (isatin) derivatives and nalidixic acid carbohydrazide, synthesis, antitubercular activity and pharmacophoric model building, Eur. J. Med. Chem. 45 (2010) 4578-4586.

- [22] A. Shinde, S. Zangade, S. Chavan, Y. Vibhute, Microwave induced synthesis of bis-Schiff bases from propane-1,3-diamine as promising antimicrobial analogs, Org. Commun. 7 (2014) 60-67.
- [23] M. Hranjec, K. Starcevic, S.K. Pavelic, P. Lucin, K. Pavelic, G.K. Zamola, Synthesis, spectroscopic characterization and antiproliferative evaluation in vitro of novel Schiff bases related to benzimidazoles, Eur. J. Med. Chem. 46 (2011) 2274-2279.
- [24] N.V.S. Rao, T.D. Choudhury, R. Deb, M.K. Paul, T.R. Rao, T. Francis, I.I. Smalyukh, Fluorescent lanthanide complexes of Schiff base ligands possessing N-aryl

moiety: influence of chain length on crossover (calamitic to discotic) phase behaviour, Liq. Crys. 37 (2010) 1393-1410.

- [25] A. Guha, J. Adhikary, T. Mondal, D. Das, Zinc and cadmium complexes of a Schiff base ligand derived from diaminomaleonitrile and salicylaldehyde: Syntheses, characterization, photoluminescence properties and DFT study, Ind. J. Chem. 50A (2011) 1463-1468.
- [26] G. Consiglio, S. Failla, P. Finocchiaro, I.P. Oliveri, S.D. Bella, Aggregation properties of bis(salicylaldiminato) zinc(II) Schiff-base complexes and their Lewis acidic character, Dalton Trans. 41 (2012) 387-395.