Effect of Rotational Speed on Linear and Non-Linear Optical Properties of Sol-Gel Spin Coated Nanostructured CdS Thin Films

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ABSTRACT

Cadmium sulfide thin films have been prepared by sol-gel spin coating method on glass substrates at different rotational speeds. Glancing angle X-ray diffraction pattern of CdS thin films exhibit cubic phase with (1 1 1) as preferential orientation. The surface morphology of the CdS thin films was studied by atomic force microscope. The electrical resistivity of CdS thin films decreased from 5.66 × 1010 to 1.72 × 108 Ω.cm with the increase of rotational speed from 1500 to 3000 rpm. The optical transmittance of CdS thin films increased because of reduction in film thickness with the increase of rotational speed. The optical band gap of CdS thin films decreased from 2.65 to 2.46 eV with the increase of rotational speed due to quantum confinement effect. The normal dispersion of refractive index for CdS thin films is described using Wemple-DiDomenico single-oscillator model. Optical dispersion parameters such as single-oscillator energy (E0), the dispersion energy (E0) and the static refractive index (n0) were calculated. The Verdet coefficient (V) is determined based on the refractive index dispersion study. The first and third order nonlinear optical susceptibility (χ(2), χ(3)) and nonlinear refractive index (n2) are determined for CdS thin films.

1. Introduction

Cadmium sulfide (CdS) belongs to II-VI semiconducting material with a direct band gap of 2.42 eV (in cubic phase) and 2.57 eV (in hexagonal phase) at room temperature. CdS exhibit in two forms namely α-CdS (hexagonal wurtzite, space group: P63mc) and β-CdS (cubic zinc blende crystal structure, space group: F43m). β-CdS forms at low temperature with a metastable phase where as α-CdS forms at high temperatures with a stable phase; however, annealing at high temperatures β-CdS can transform to α-CdS [1]. CdS had a bulk refractive index of 2.52 at wavelength 600 nm which is well suited for solar cell applications [2]. Cadmium sulfide (CdS) is most widely used in heterojunction thin-film photovoltaics as a window layer in CdTe thin-film solar cells and as a buffer layer in Cd(In,Ga)Se2 (CIGS) thin-film solar cells. CdS most widely found applied material in electro-optic devices such as photoconducting cells, photosensors, transducers, optical wave-guides and non-linear integrated optical devices, diluted magnetic semiconductors in spintronic devices, light emitting diodes, field effect transistors and gas sensors [3-7]. CdS thin films have been prepared by different workers using different methods which includes pulsed laser deposition (PLD) [8], sputtering [9], thermal evaporation [10], metal organic chemical vapor deposition (MOCVD) [11], chemical bath deposition (CBD) [12], layer adsorption and reaction (SILAR) [13], spray pyrolysis [14] and sol-gel route [15]. Each method produces the films with different physical properties, which should be optimized for specific applications. Among these methods, the sol-gel spin coating method is a simple, low cost, and ability to produce uniform films with good adherence, provides a high specific surface area, narrow pore size, better microstructural control of metallic particles and uniform particle distribution [16]. Many researchers reported the structural, electrical and optical properties of CdS thin films. Studies on the nonlinear optical properties of sol-gel spin coated CdS thin films are very rare to find in the literature. The main objective of this paper is to study the effect of rotational speed on the linear and non-linear optical properties of CdS thin films and to report the detailed information on optical constants, optical dispersion parameters and non-linear optical parameters which are of considerable importance for its applications in integrated optical devices such as switches, filter, modulators, etc.

2. Experimental Methods

CdS thin films have been prepared by mixing 0.6 mL of polyethylene glycol (PEG 200) with 8.9 mL of ethanol and 0.5 mL of acetic acid under stirring continuously for 90 minutes. Cadmium nitrate (Cd(NO3)2·4H2O) and thiourea (SC(NH2)2) chemicals of analytical grade from M/S Fine Chem. Pvt. Ltd., Mumbai, India) were used as precursors for cadmium and sulfur source. These precursors were dissolved in an ethanol with stirring for 90 minutes. The as-prepared solution was slowly added to PEG sol with stirring for 6 hours in order to obtain the final sol, which is used for depositing CdS thin films. The spin coating method was used to deposit CdS thin films using the above sol on glass substrates. The glass substrates were cleaned thoroughly with detergent, ethanol and acetone for 10 min. Finally, the substrates were dried under an infra-red (IR) lamp. These glass substrates were used for the preparation of CdS thin films. CdS thin films were prepared on glass substrates by spin coating method with varying rotational speed from 1500 to 3000 rpm and annealed in vacuum at 200 °C. The thickness of the CdS thin films was measured by using Talysurf thickness profilometer.

Structural properties of CdS thin films were analyzed by Glancing angle X-ray diffraction (GAXRD) using a Rigaku X-ray diffractometer with a CuKα radiation (λ=0.154 nm) source operated at 40 kV and 40 mA. X-ray diffraction measurements were recorded in the 2θ range of 20°-90° at a glancing angle of 2° with a scan speed of 1° min-1. The surface morphology of the CdS thin films was studied using atomic force microscope (AFM; Model: Park NX20). The thickness of the CdS films is measured by Dektak surface profilometer. The electrical measurements of CdS thin films were carried out by four-probe method and Hall effect setup. The optical transmittance spectra of the CdS thin films were recorded by using UV-Vis/NIR spectrometer (Shimadzu MPC3600) in the wavelength range of 300-2500 nm.
3. Results and Discussion

3.1 Structural Analysis

Glancing angle X-ray diffraction (GAXRD) patterns of CdS thin films prepared by sol-gel spin coating method on glass substrates by varying rotational speeds from 1500 to 3000 rpm are shown in Fig. 1. The diffraction peaks for CdS thin films are indexed to (1 1 1), (2 0 0) and (3 1 1) planes of pure CdS [JCPDS Card No. 10-0454]. It is noticed that all the films exhibit (1 1 1) preferred orientation with a cubic structure.

\[ \theta = \sin^{-1} \left( \frac{a}{2} \right) \]

where \( \theta \) is Bragg angle, \( \lambda \) is the wavelength of X-ray (\( \lambda = 0.154 \text{ nm} \)) and \( n \) is an integer. The interplanar spacing \( d \) of CdS thin films is calculated using the Bragg’s equation,

\[ a = d \left( h^2 + k^2 + l^2 \right)^{1/2} \]

For the observed \( d \) values corresponding to \((h k l) \) plane obtained by X-ray diffraction data. The lattice constant \((a) \) values increases from 0.5813 to 0.5826 nm with the increase of rotational speed from 1500 to 3000 rpm. The increase of interplanar spacing and lattice constant with the increase of rotational speed may be due to the improvement in crystallinity of the films. The number of crystallites per unit area \((N) \) is calculated using the formula,

\[ N = \frac{I}{D} \]

where \( I \) is the film thickness and \( D \) is the crystallite size. The thickness of the films measured from surface profilometer decreased from 320 to 180 nm with the increase of rotational speed. It is due to the fact that an increase in rotational speed causes the centrifugal force on the deposited solution on the substrate which makes the film thin. The number of crystallites \((N) \) of CdS thin films decreases from 11.08 × 10^{15} to 2.78 × 10^{15} m^{-2} with the increase of rotational speed. The structural parameters of CdS thin films post-annealed at different temperatures are shown in Table 1.

3.2 AFM Analysis

Fig. 2 shows the AFM images (scan area of 2 μm × 2 μm) for CdS thin films taken in non-contact mode. From AFM images, it is observed that the grain size increases with the increase of rotational speed due to the increase in grain boundary movement and mobility of the thin films. The average roughness \((R_a) \) of the CdS thin films increased from 0.92 to 2.34 nm, whereas the RMS roughness \((R_q) \) values increased from 6.6 to 18.5 nm with the increase of rotational speed from 1500 to 3000 rpm. The increase in surface roughness of the films decreases the reflective loss which consequently increases the quantum efficiency of a solar cell [18]. The maximum peak height \((R_p) \) and peak valley \((R_{pv}) \) values increased with the increase of rotational speed due to the improvement in crystallinity of the films. The negative values of skewness \((R_s) \) indicate that the film had a rough surface with a low density of dips or valleys and flat regions. When kurtosis \((R_k) \) value is equal to 3, it represents the surface with a Gaussian height distribution and the surface is called Mesokurtic, but if \( R_k > 3 \), the surface is flat and called platykurtic. If \( R_k < 3 \), the surface has more number of peaks than valleys. The CdS thin films exhibit the value of \( R_k > 3 \) and the distribution will have a relatively higher number of peaks and low valleys with a spiky surface. The statistical analysis of CdS thin films obtained from AFM is given in Table 2.

3.3 Electrical Properties

The electrical resistivity \((\rho) \) of the CdS thin films were measured by using a four-probe method. The electrical resistivity of CdS thin films decreased from 5.66 × 10^{10} to 1.72 × 10^{10} Ω cm with the increase of rotational speed from 1500 to 3000 rpm. The decrease in electrical resistivity of the CdS thin films is due to the decrease in the density of grain boundary intercrystallines and defects such as pinholes, voids etc. The obtained electrical resistivity values are in good agreement with previously reported works [19, 20]. The carrier concentration \((n) \) and mobility \((\mu) \) of the CdS thin films were obtained from Hall effect measurement at room temperature in a magnetic field of 0.35T [21]. The sign of Hall voltage across the samples is negative, which indicates an n-type semiconducting behavior of the films. The carrier concentration of CdS thin films increased from 0.88 × 10^{19} to 1.36 × 10^{20} cm^{-3} with the increase of rotational speed from 1500 to 3000 rpm. The increase in the carrier concentration of the films causes shrinkage in the gap known as the bandgap narrowing, which signifies that the density of dislocations and density of grain boundaries will be decreased [22]. The carrier concentration is in the order of 10^{22} cm^{-3} for all the films and these values are in good agreement with chemical bath deposited CdS thin films [23, 24]. The Hall mobility of the films increased from 125 to 265 cm^2 V^{-1} s^{-1} with the increase of rotational speed from 1500 to 3000 rpm due to the increase in the grain size and reducing in density of grain boundary intercrystallines and defects.
grain boundary scattering. The mean free path \( L \) of an electron is calculated using the equation [25].

\[
L = \left( \frac{3\pi^2}{\lambda^2} \right) \left( \frac{h}{2\pi e^2} \right), \quad 1/\rho \cdot \eta n_1^{1/3}
\]

(4)

where \( h \) is Planck’s constant, \( e \) is the electron charge, \( \rho \) is the resistivity, and \( n_1 \) is the carrier concentration. The calculated mean free path \( L \) of an electron of CdS thin films varies from 4.76 to 11.70 nm. The obtained values of mean free path are much shorter than the grain size measured according to XRD analysis. They reported that the effect of scattering at grain boundaries and dislocations can be neglected when the mean free path is smaller than the grain size, and concluded that scattering of conduction electrons depends on neutral and ionized impurities. The electrical properties of CdS thin films prepared with different rotational speeds are given in Table 3.

<table>
<thead>
<tr>
<th>Rotational speed (rpm)</th>
<th>Type</th>
<th>Electrical Carrier Resistivity, ( \rho ) (( \Omega \cdot \text{cm} ))</th>
<th>Mobility, ( \mu ) (( \text{cm}^2 / \text{V} \cdot \text{s} ))</th>
<th>Mean free path, ( L ) (nm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1500</td>
<td>n</td>
<td>5.66 x 10^10</td>
<td>125</td>
<td>4.76</td>
</tr>
<tr>
<td>2000</td>
<td>n</td>
<td>4.64 x 10^10</td>
<td>141</td>
<td>5.52</td>
</tr>
<tr>
<td>2500</td>
<td>n</td>
<td>3.30 x 10^10</td>
<td>167</td>
<td>6.94</td>
</tr>
<tr>
<td>3000</td>
<td>n</td>
<td>1.72 x 10^10</td>
<td>265</td>
<td>11.70</td>
</tr>
</tbody>
</table>

3.4.1 Optical Properties

3.4.1.1 Optical Transmittance and Energy Band Gap

The optical transmittance spectra of CdS thin films prepared with different rotational speeds are shown in Fig. 3(a). The shift of absorption edge towards longer wavelengths with an increase in rotational speed indicates the growth of the particles on annealing. The average optical transmittance of CdS thin films [14] in the UV-Vis region is increased from 72 to 90% with the increase of rotational speed which may be caused by lattice imperfections, reduction in voids, film thickness and increase in the crystallite size. Furthermore, the films revealed high transparency in the visible region, \( T(\lambda) > 70\% \) makes useful as a transparent electrode for optoelectronic devices. Fig. 3(b) shows the extrapolation of \( (\alpha v)^2 \) versus \( h \nu \) with the intercept of the straight line on \( h \nu \) axis the optical band gap of CdS thin films. The optical band gap of CdS thin films decreased from 2.65 to 2.46 eV with the increase of rotational speed from 1500 to 3000 rpm due to quantum size confinement effect. The obtained direct band gap values are in good agreement with the previously reported values for cubic CdS thin films [26, 27]. The optical constants of CdS thin films were calculated from the equations as reported in the earlier literature [28, 29]. The absorption coefficient \( \alpha \) values of CdS thin films prepared at different rotational speeds are in the range of \( 1.23 \times 10^4 – 4.87 \times 10^4 \) cm\(^{-1}\). The variation of extinction coefficient \( (k) \) for CdS thin films as a function of wavelength is shown in Fig. 3(c).

![Fig. 3](https://doi.org/10.30799/j.ast.238.19050210)

Fig. 3 Variation of (a) Optical transmittance spectra versus wavelength; (b) Tauc’s plot for CdS thin films; (c) Extinction coefficient, \( k \) and (d) Refractive index, \( n \) for CdS thin films as a function of wavelength

The extinction coefficient values are in the range of 0.046 – 0.155, whose values decreases with the increase of wavelength. The low values of the extinction coefficient for the CdS films in the visible and near infrared region show their surface smoothness and high optical transmittance. The optical band gap of the CdS thin films was determined by using the Tauc’s formula [30, 31]. The variation of refractive index \( (n) \) with wavelength for CdS thin films prepared with different rotational speeds is shown in Fig. 3(d). The refractive index \( (n) \) of the films decreased from 2.78 to 2.42 (at wavelength 600 nm) with the increase of rotational speed from 1500 to 3000 rpm due to the improvement in crystallinity of the films. At the shorter wavelengths, the refractive index value increases rapidly and at energies lower than absorption edge, its value falls down and saturates at higher wavelength representing the normal dispersion behaviour which can be explained in terms of single oscillator model. It is also clear that, the effect of rotational speed on the refractive index and semiconductor dispersion profiles exhibit a linear displacement in the dispersion profile with decreasing refractive index. The refractive index of the films declines towards longer wavelengths which may be due to the influence of lattice absorption. H. Metin et al. [32] reported similar behavior in the change of refractive index for the CdS thin films prepared by the chemical deposition method.

### 3.4.2 Optical Dispersion, Group and Phase Velocity

The optical dispersion is an important parameter to understand the optical properties of the films. Fig. 4(a)-(c) shows the optical dispersion, group and phase velocity of films with photon wavelength respectively. It is seen that optical dispersion \( dn/d\lambda \) of the film depends on both rotational speed and wavelength. The group velocity, \( U_g \) describes pulse propagation through an isotropic medium with undistorted shape is expressed as [33].

\[
U_g = \frac{d\nu}{dk} = c \left( \frac{n - \lambda}{\lambda} \right)
\]

(5)

where \( c \) is the velocity of light in vacuum, \( k \) is the wave number, \( \omega \) is the angular frequency of the incident light. The wavelength of dependence of the group velocity and phase velocity for the CdS thin film is shown in Fig. 4(b) and 4(c). The results show that the group velocity and phase velocity for CdS thin films increases with wavelength, but decreases with rotational speed which may be due to the optical dispersion in the material.

![Fig. 4](https://doi.org/10.30799/j.ast.238.19050210)

Fig. 4 Variation of (a) optical dispersion, \( dn/d\lambda \); (b) group velocity, \( U_g \) and (c) phase velocity, \( v_p/c \) for CdS thin films as a function of wavelength

### 3.4.3 Magneto-Optical Constant: Verdet Coefficient (V)

The Verdet coefficient \( (V) \) is known as the single pass volume Faraday rotation per unit magnetic field and unit thickness. The Verdet coefficient is normally measured using two Glan-Taylor polarizers, but in the present study we estimated it from the refractive index dispersion [34]. The refractive index is related to \( V \) as follows.

\[
V(\lambda) = \mu_s r (e/2mc) \lambda dn/d\lambda
\]

(6)

where \( \mu_s \) is the vacuum permeability, \( r \) is the magneto-optical anomaly factor which is close to 0.28 for materials exhibiting covalent bond character and close to 1 for materials showing ionic type bond, \( e \) is the charge of electron, \( m \) is the mass of electron, \( c \) is the velocity of light and \( dn/d\lambda \) is the optical dispersion of the material. The wavelength dependence of the Verdet coefficient for CdS thin films prepared with different rotational speeds is shown in Fig. 5. It is observed that the Verdet coefficient takes positive and negative values and mainly depends on the band gap in the negative part.
The dispersion of refractive index in semiconductors has been analyzed using the concept of the single-oscillator model developed by Wemple and DiDomenico. From this model the energy parameters, single-oscillator energy ($E_0$) and dispersion energy ($E_d$) are introduced and the refractive index ($n$) at any photon energy ($\hbar\nu$) is expressed by the Wemple-DiDomenico relationship [35],

$$n^2 = 1 + \frac{E_d}{E_0 - \hbar\nu}$$  \hspace{1cm} (7)

By plotting $(n^2-1)$ versus $(h\nu)^2$ and fitting straight lines to the points as shown in Fig. 6, the oscillator parameters $E_d$ and $E_0$ are determined from the slope ($E_dE_0^{-1}$) and the intercept on vertical axis ($E_0/E_d$). The values of $E_d$ and $E_0$ are given in Table 4, it is noticed that these values decreases with the increase of rotational speed because of structural order and partial reconnection of atomic and molecular liaisons.

Fig. 6 Plot of $(n^2-1)^{-1}$ vs $(h\nu)^2$ for CdS thin films

Table 4 Optical dispersion parameters and non-linear optical parameters of CdS thin films

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Rotational speed (rpm)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>1500</td>
</tr>
<tr>
<td>$E_d$ (eV)</td>
<td>2.65</td>
</tr>
<tr>
<td>$E_0$ (eV)</td>
<td>3.08</td>
</tr>
<tr>
<td>$E_0$ (cm$^{-1}$)</td>
<td>7.71</td>
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<tr>
<td>$M_3$ (cm$^{-1}$)</td>
<td>2.50</td>
</tr>
<tr>
<td>$M_3$ (eV$^{-2}$)</td>
<td>0.26</td>
</tr>
<tr>
<td>$\tau$ (ps)</td>
<td>1.87</td>
</tr>
<tr>
<td>$c_0$ (10$^{10}$ cm$^{-3}$)</td>
<td>6.36</td>
</tr>
<tr>
<td>$N/m^*$ (10$^{10}$ cm$^{-3}$)</td>
<td>3.77</td>
</tr>
<tr>
<td>$\omega$ (10$^{13}$ s$^{-1}$)</td>
<td>3.34</td>
</tr>
<tr>
<td>$\lambda$ (10$^{13}$ Hz)</td>
<td>3.00</td>
</tr>
<tr>
<td>$\chi^{(3)}$ (esu)</td>
<td>0.20</td>
</tr>
<tr>
<td>$\chi^{(3)}$ (10$^{13}$ esu)</td>
<td>2.66</td>
</tr>
<tr>
<td>$n$ (10$^{15}$ esu)</td>
<td>5.37</td>
</tr>
</tbody>
</table>

From the analysis of $(n^2-1)^{-1}$ against $(h\nu)^2$, it allows to determine the static refractive index ($n_s$) using the relation,

$$n_s = \left(1 - \frac{E_d}{E_0}\right)^{-\frac{1}{2}}$$  \hspace{1cm} (8)

By extrapolating the Wemple-DiDomenico relation to the value of the incident photon energy $(h\nu)$ approaching zero. The calculated values of the static refractive index ($n_s$) for all the films are reported in Table 4. Using $E_d$ and $E_0$ obtained from the fitted transmittance spectra, $M_3$ and $M_3$ moments of the optical spectra are determined from the following two equations [36],

$$E_d = \frac{M_1}{M_3}, \quad E_0 = \frac{M_2}{M_3}$$  \hspace{1cm} (9)

$$E_d = \frac{M_1}{M_3}, \quad E_0 = \frac{M_2}{M_3}$$  \hspace{1cm} (10)

These optical moments represent the measure of the average bond strength and single-oscillator approximation to the dielectric response of the material. The two moments $M_1$ and $M_2$ were calculated from the data of $E_d$ and $E_0$ are given in Table 4.

3.4.5 Optical Dielectric Constant and Optical Conductivity

The complex dielectric constant parameter provides the electronic structure of the material and helps in the design of highly efficient optoelectronic devices. The dependence of the refractive index ($n$) on the real dielectric constant ($\varepsilon_r$) is given by the Spitzer-Fan model [37],

$$\varepsilon_r = n^2 \left(1 - \frac{\varepsilon_0}{\varepsilon_0 + \frac{N}{m^*}}\right)^{\frac{1}{2}}$$  \hspace{1cm} (11)

where $\varepsilon_0$ is the high-frequency dielectric constant, $e$ is the electronic charge, $c$ is the speed of light, $N/m^*$ is the ratio of carrier concentration to the effective mass, $\omega_0$ is the plasma frequency and $\tau$ is the optical relaxation time. The plot of real dielectric constant ($\varepsilon_r$) versus $\lambda^2$ for CdS thin films prepared with different rotational speeds are shown in Fig. 7(a). It is observed that the dependence of $\varepsilon_r$ on $\lambda^2$ is linear at longer wavelengths, the values of $\varepsilon_r$ and $N/m^*$ were deduced from the extrapolation of these plots to $\lambda^2=0$ and from the slope of the graphs. Fig. 7(b) shows the variation of imaginary dielectric constant ($\varepsilon_i$) with $\lambda^2$ for CdS thin films, it is found that $\varepsilon_i$ is lower than $\varepsilon_r$ which proves that the energy loss of light through thin films is low. Generally, the variation of dielectric constant with the rotational speed is attributed due to the defects such as stresses, voids, inhomogeneity, discontinuities, grain boundaries, etc. The real and imaginary part of dielectric constant of CdS thin films decreases with the increase of rotational speed from 1500 to 3000 due to the grain boundary defects. Similar trend in the variation of real and imaginary dielectric constant of CdS thin films prepared with different rotational speed is noticed, it is due to loss of energy which occurs due to the absorption of energy by free carriers present in the medium. It is noticed that the carrier concentration increases, band gap narrowing occurs due to many body effects and shifting of band edge towards the lower energy.

Fig. 7 Variation of (a) real part of the dielectric constant ($\varepsilon_r$) versus $\lambda^2$ and (b) imaginary part of the dielectric constant ($\varepsilon_i$) versus $\lambda^2$ for CdS thin films

The plasma frequency ($\omega_0$) is related to the free carrier concentration to effective mass ratio ($N/m^*$) from the following relation,

$$\omega_0^2 = \frac{4 \pi N e^2}{\varepsilon_r m^*}$$  \hspace{1cm} (13)

From the $\omega_0$, we have determined the relaxation time $\tau$ whose values are given in Table 4. It is observed that both $\varepsilon_r$ and $N/m^*$ values increased with increasing rotational rate from 1500 to 3000 rpm. The plasma frequency evaluated using Eq.(13) increases with increasing rotational speed (Table 4) which may be due to the increase in free carrier concentration.

The optical conductivity ($\sigma_{opt}$) describes the dissipation of electromagnetic energy in the material and proportional to absorption energy as,

$$\sigma_{opt} = \text{Im}(\varepsilon) / 4\pi$$  \hspace{1cm} (14)
where $a$ is the absorption coefficient, $n$ is the refractive index and $c$ is the velocity of light. Fig. 8 shows the variation of optical conductivity $\sigma_{\omega}$ as a function of photon energy $h\nu$ for CdS thin films. The values of $\sigma_{\omega}$ are found to be in the range of $1.8 \times 10^{-3}$ to $1.77 \times 10^{15}$ m$^{-1}$ s for the CdS thin films prepared with rotational speed from 1500 to 3000 rpm. The increase in optical conductivity at high photon energies is due to the electron excited by photon energy. The overall high values of $\sigma_{\omega}$ for all films increase the photo response behavior and make them suitable for optoelectronic applications.

4. Conclusion

CdS thin films were prepared on glass substrates by sol-gel spin coating method with rotational speed of 1500, 2000, 2500 and 3000 rpm. All the films exhibit cubic structure with preferential growth along the (111) plane confirmed by XRD analysis. The electrical resistivity of the films was found to be decreased with the increase of rotational speed. The optical transmittance of CdS thin films increases with the increase of rotational speed because of decrease in film thickness. The optical band gap of CdS thin films decreased from 2.65 to 2.46 eV with the increase of rotational speed. The Verdet coefficient for the CdS thin films was calculated from the refractive index dispersion and it exhibited a spread negative range, which was dependent on the band gap energy. The dispersion of the refractive index of CdS thin films was analyzed by applying the Wemple-DiDomenico single oscillator model. It was found that the oscillator resonance energy $E_o$ and oscillator dispersion energy $E_d$ decreased with increasing rotational speed. In addition, nonlinear optical parameters like linear optical susceptibility $\chi^{(1)}$, nonlinear optical susceptibility $\chi^{(3)}$, and nonlinear refractive index $n_2$ are also reported. The results in this work revealed a possibility of producing high-quality CdS thin films for future nonlinear optical devices/ optoelectronic devices.

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